

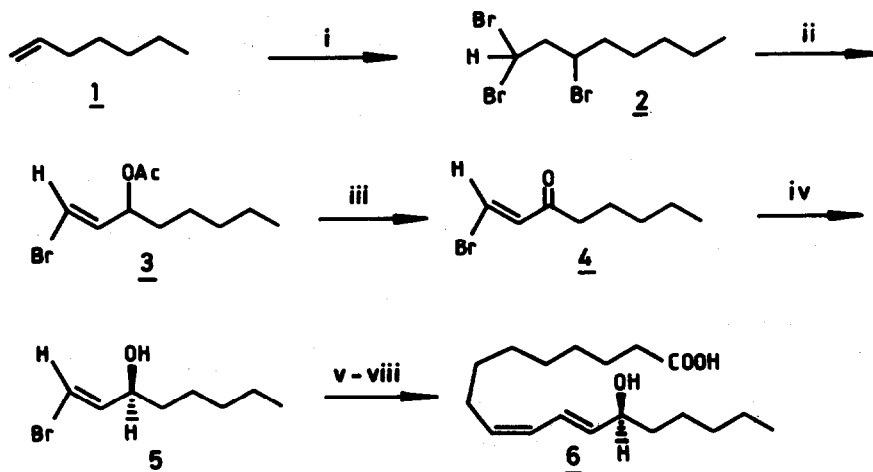
**A NOVEL CHEMO-ENZYMATIC ENANTIOSPECIFIC SYNTHESIS OF (S)-CORIOLIC ACID
MEDIATED VIA IMMOBILIZED ALCOHOL DEHYDROGENASE OF BAKER'S YEAST**

U.T. Bhalarao^{*}, L. Dasaradhi, C. Muralikrishna and N.W. Fadnavis

Indian Institute of Chemical Technology, Hyderabad 500 007, India.

Abstract: (S)-(E)-(+)-1-Bromo-1-octene-3-ol was prepared in 85% yield and high optical purity (e.e. 97.4%) via a stereospecific reduction of corresponding bromo vinyl ketone using alcohol dehydrogenase from baker's yeast (EC 1.1.1.1) and NAD(P)H immobilized on Nucleosil 120-5 C₁₈, and converted to 13(S)-hydroxy-9Z,11E-octadecadienoic (coriolic) acid.

13(S)-Hydroxy-9Z,11E-octadecadienoic (coriolic) acid **6** is an oxygenated unsaturated fatty acid formed as a product of metabolism of linoleic acid in plants and animals¹, and possesses interesting biological activities². Recently its synthesis has attracted considerable attention³ as it was found to act as a self defence substance in rice plants⁴. Herein we report a short and simple enantiospecific synthesis of **6** following the pathway depicted in scheme I. The chiral bromo vinyl alcohol **5** is the key target in the synthesis of (S)-coriolic acid. However, reduction of α,β -unsaturated ketones such as **4** with whole cells to alcohol **5** is difficult⁵ and mostly leads to reduction of double bond instead of keto function or reduction of both^{3a}. We found that enzyme alcohol dehydrogenase from baker's yeast (E.C. 1.1.1.1) reduces **4** enantioselectively to (S)-**5** keeping the bromo vinyl group intact in presence of NAD(P)H as cofactor.



SCHEME - I

Scheme I: i) Cu, CHBr₃; ii) KOAc (2 eq.), 18-crown-6, DMF; iii) K₂CO₃/MeOH, then PCC; iv) ADH from baker's yeast and NAD(P)H immobilized on Nucleosil 120-5 C₁₈, ethyl acetate-isooctane 1:9, isopropanol, pH 7.5; v) H-C≡C-(CH₂)₇-COOMe, Pd(PPh₃)₄, CuI, PrⁿNH₂; vi) H₂-Pd; BaCO₃; vii) K₂CO₃-MeOH; viii) H⁺.

Our attempts to use Klivanov's methodology of enzyme immobilized on glass beads - substrate in ethyl acetate were not successful, prompting us to look for an alternative methodology. We have successfully exploited the property of Nucleosil C18 of absorbing both hydrophilic as well as hydrophobic compounds to bring together the enzyme, the coenzyme NAD(P)H and the highly hydrophobic substrate. Recycle of the oxidized coenzyme NAD(P) is achieved by *in situ* enzymatic oxidation of isopropanol to acetone and consequent reduction of NAD(P) to NAD(P)H.

Copper catalysed bromoform addition to commercially available 1-heptene 1 gave 3-bromo-1,1-dibromooctane 2 (82%) which on treatment with two equivalents of potassium acetate 1g-crown-6 in dimethylformamide resulted in 3-acetoxy-1-bromo-1-octene 3 (85%). Hydrolysis of 3 in methanol-potassium carbonate and oxidation with pyridinium chlorochromate (PCC) gave exclusively (E)-ketovinylbromide 4 (75%). This was enantiospecifically reduced to (S) (E)-(+)-1-bromo-1-octen-3-ol 5, (85%) with 97.4% e.e. as determined by its Mosher's ester. 5 was coupled with methyl 9-decynoate in presence of Pd(PPh₃)₄, CuI and n-propyl amine. The product (58%) on partial hydrogenation over palladium on barium carbonate furnished exclusively cis-hydrogenated 9E,11Z methyl ester (80%) which was hydrolysed to (S)-coriolic acid 10 6 (85%).

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- In a typical experiment the enzyme (5 mg, Sigma, USA) and NAD(P)H (15 mg, 20 μmoles, Spectrochem, India) were dissolved in iris buffer (5 mL, 0.05 M, pH 7.5) and mixed with Nucleosil 120-5 C₁₈ (3g). The slurry was air dried for 6-8 hrs and stirred with solution of 1 (205 mg, 1 mmole) in ethyl acetate-isooctane (1:9, 50 mL) at r.t. Isopropanol (4 mL) was added in aliquots of 200 μL over 5 days. The reduction was complete in one week as followed by t.l.c. Filtration and column chromatography yielded 5 (175 mg, 85%) with e.e. 97.4% as determined by its Mosher's ester.
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- ¹H NMR (200 MHz, CDCl₃): 4.0-0.91 (t, 3H, Me), 1.2-1.6 (br, s, 6H, 3CH₂) 2.50 (t, 2H, COCH₂), 6.80 (d, 3H, I-I, IH=CHBr), 7.55 (d, 3H, Hz, IH=CHCO), 5.0-12.7° (c 1.39, metha, e.e. 97.4%, ii {1-} for (5) 5 + 13.1° (c 1.39, CHCl₃), -0.92 (t, 3=7.2Hz, 3H, Me), 1.25-1.75 (br, s, 4H, 4CH₂), 1.94 (s, IH, OH), 4.12 (~, IH, CHOH), 6.0-6.4 (rrb 2H, -CH), 61"~9.1 (c 1.29 • CHCl₃), lit. ~, ~+9.3 (c 1.29 CHCl₃). ¹H NMR (200 MHz, CDCl₃) 0.91 (t, 3H, 3-7 2Hz Me), 1.05-1.75 (br, s, 1-gH, gCH₂), 1.91-2.12f (rn, 2H, CH~CH=CH), 2.32 (t, ~H, C~oCOOH), 4.12 (rrb 1H, C~[OH]), 4.71 (br s, IH, OH), 5.13-6.15 (rrb 3H, C~I=CH-CH=CH-CHOH), 6.42 (dd, 3=9.g and 15Hz, IH, =CH), 9.7g (br s, IH, COOH).